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13. All acid—base reactions have two conjugate acid—base pairs. One conjugate acid—base pair in the reaction in Model 3 is  $\text{H}_2\text{O}/\text{H}^+\text{O}^-$ . List the other acid—base pair in the reaction. 14. Why is  $\text{HCO}_3^-$  considered the "acid" part of the pair in the reaction in Model 3? 15. is  $\text{CO}_3^{2-}$  considered the "base" part of the pair in the reaction in Model 3? 16.

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Model 3 — Conjugate Acid—Base Pairs Base + 1-120 (l) HC031-(aq) 13. All acid—base reactions have two conjugate acid—base pairs. One conjugate acid—base pair in the reaction in Model 3 is H O<sup>+</sup>/H O. List the other acid—base pair in the reaction. 14. Why is FICO - considered the "acid" part of the pair in the reaction in Model 3? 15.

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Chem 116 POGIL Worksheet - Week 8 - Solutions Equilibrium Continued Introduction to Acid-Base Concepts ... Check your final answers with a Q calculation. Let x be the number of mol/L of  $\text{H}_2$  or  $\text{I}_2$  ... acid 1 base 2 base 1 acid 2  $\text{CH}_3\text{NH}_2(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{CH}_3\text{NH}_3^+(\text{aq}) + \text{OH}^-(\text{aq})$  base 1 acid 2

## **Chem 116 POGIL Worksheet - Week 8 - Solutions Equilibrium ...**

Naming Acids and Bases Name the following acids and bases: 3) 6) 7)  $\text{NaOH}$   $\text{H}_2\text{SO}_3$   $\text{H}_3\text{PO}_4$   $\text{NH}_3$  acid  
hos Write the formulas of the following acids and bases:

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