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IDEAL SOLUTIONS; The solutions which obeys the Raoult's Law over the entire range of concentration are known as Ideal Solution. PROPERTIES : (i)  $\Delta_{\text{mix}} H = 0$  i.e. The enthalpy of mixing of the pure component to form solution is 0. It is neither exothermic nor endothermic . (ii)  $\Delta_{\text{mix}} V = 0$  i.e. There is no change in volume when 2 components are mixed to form solution.

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