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following questions in the space
provided. 1. Match description on the
right to the correct crystal type on the

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left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting.

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Theory of Gases 1- Gases consist of large numbers of tiny particles that are far apart relative to their size. 2- Collisions between gas particles and between particles and their container walls are elastic.

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SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. a When a substance in a closed system undergoes a phase change and the system reaches equilibrium, (a) the two opposing changes occur at equal rates. (b) there are no more phase changes. (c) one phase change predominates.

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States of Matter SECTION 1 SHORT

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ANSWER Answer the following questions in the space provided. 1. Identify whether the descriptions below describe an ideal gas or a real gas. ideal gas

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organizer(s) for this chapter LOOK AT MIDTERM EXAM REVIEW AND ADD NOTES FROM IT TO ALL CHAPTER REVIEW PAGES The following pages contain the bulk (but not all) of the information for the chapter 1 test. Focus on this content, but make sure to review class notes, activities ...

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